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FOURTH SEMESTER M.Sc. DEGREE (REGULAR/SUPPLEMENTARY) EXAMINATION, APRIL 2024

(CBCSS)

Physics

PHY 4C 12—ATOMIC AND MOLECULAR SPECTROSCOPY

(2019 Admission onwards)

Time: Three Hours

Maximum: 30 Weightage

Section A

8 short questions, each answerable within 7.5 minutes.

Answer all questions.

Each question carries weightage 1.

- Define Shark effect.
- 2. Figure the normal modes of water molecule.
- 3. Draw the schematic diagram of IR spectrometer.
- 4. What are the selection rules for the parallel vibrations?
- 5. What is the principle of ESR?
- Define nuclear magneton.
- 7. What is the principle of FTIR?
- 8. What is chemical shift?

 $(8 \times 1 = 8 \text{ weightage})$

Section B

4 Essay questions, each answerable within 30 minutes.

Answer any two questions,

Each question carries weightage 5.

- 9. Explain rotational fine structure of electronic vibration spectra. Explain P,Q and R branches Explanation, figure.
- 10. Explain vector atom model. With necessary figure briefly explain:
 - (a) L-S coupling.
 - (b) J-J coupling?

Turn over

- 11. Explain diatomic vibrating rotator. Explain with figure Vibrational Raman spectra.

 $(2 \times 5 = 10_{\text{Weigh}})$

Section C

7 problem questions, each answerable within 15 minutes.

Answer any four questions.

Each question carries weightage 3.

- 13. Rotational and centrifugal distortion constants of HCl molecule are 10.593 cm 5.3×10^{-4} cm⁻¹ respectively. Estimate the vibrational frequency and force constant of the m₀
- Explain dissociation.
- 15. Explain with figure vibrational coarse structure.
- 16. The band origin of a transition in C2 is observed at 19,378 cm⁻¹ while rotational structure indicates that the rotational constants in excited and ground states are respect $B' = 1.7527 \text{ cm}^{-1}$ and $B' = 1.6326 \text{ cm}^{-1}$ Estimate the position of the band head
- 17. Derive the classical theory of Raman scattering.
- 18. Draw and explain Fortrat parabola.
- 19. Explain nuclear magnetic resonance condition.

 $(4 \times 3 = 12 \text{ weight})$