

## FOURTH SEMESTER M.Sc. DEGREE EXAMINATION, MARCH 2020

(CUCSS)

Chemistry

CH 4C 12—INSTRUMENTAL METHODS OF ANALYSIS

Time : Three Hours

Maximum : 36 Weightage

## Section A

*Answer all questions.**Each question carries a weightage of 1.*

1. Distinguish between accuracy and precision.
2. What is student 't' test ?
3. Explain with example colour change interval in acid-base titration.
4. In the titration of  $\text{Fe}^{2+}$  against  $\text{KMnO}_4$  in acid medium,  $\text{Cl}^-$  ions should be avoided. Why ?
5. What are the advantages of dropping mercury electrode in polarography ?
6. What is differential pulse polarography ?
7. Explain with example biamperometry.
8. Name two sources used in IR spectrometers.
9. What is nebulization ?
10. Distinguish between electron spectroscopy and electronic spectroscopy.
11. What is the experimental parameter measured in (a) DTG ; (b) DSL ?
12. What is meant by 'size exclusion chromatography' ?

 $(12 \times 1 = 12 \text{ weightage})$ 

## Section B

*Answer any eight questions.**Each question carries a weightage of 2.*

13. The following data were obtained for x-y plot. Find the slope and intercept by linear least square method :

$x$	:	0.001	0.002	0.003	0.004	0.005	0.006
$y$	:	20.60	20.72	20.82	20.93	21.03	21.15

Turn over



14. The solubility of a sparingly soluble salt  $A_2X_3$  is  $5 \times 10^{-5}$  moles per litre. Find the solubility product. What would be the solubility of the salt in  $10^{-2}$  M solution of  $X^{3-}$ ?
15. Draw the potential-volume curve for a redox titration with the help of one example. Discuss.
16. Discuss the working of a biocatalytic electrode.
17. Briefly discuss chronopotentiometry as an analytical technique.
18. Discuss the advantages of fluoremetry over *uv*-visible spectrophotometry.
19. Briefly discuss instrumentation in XRF.
20. Discuss one method of measuring gamma radiation.
21. How would you select column for GC? Discuss.
22. Briefly discuss ion exchange chromatography.
23. How would you select indicator for nonaqueous titration? Explain.
24. With the help of one example discuss organic polarography.

(8 × 2 = 16 weightage)

### Section C

*Answer any two questions.  
Each question carries a weightage of 4.*

25. Discuss instrumentation in AAS.
26. What is neutron activation analysis? Discuss.
27. Discuss theory and applications of HPLC.
28. Write a brief account of organic precipitating agents in gravimetry.

(2 × 4 = 8 weightage)