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FOURTH SEMESTER M.Sc. DEGREE EXAMINATION, MARCH 2020 (CUCSS)

Chemistry

CH 4C 13—ADVANCED TOPICS IN CHEMISTRY

Time: Three Hours

Maximum: 36 Weightage

Part A

Answer all questions. Each question carries 1 weightage.

- 1. How the properties of nanomaterials differ in macro and microstructures?
- 2. Explain briefly the principle of STM.
- 3. Which are the important criterions to select solvents in microwave induced organic reactions?
 Give few examples of solvents.
- 4. Write down Hartree's and Hartree Fock's proposal for molecular trial wave function.
- 5. Write down the Z-matrix of formaldehyde.
- 6. Distinguish between 'preorganization' and 'reorganization' in molecular recognition.
- 7. Briefly explain major classes of drugs.
- 8. Differentiate between SAR and QSAR.
- 9. Explain briefly Haughton's tea bag procedure.
- 10. Explain precipitation and co-precipitation method for the synthesis of catalysts.
- 11. What is mercury intrusion method?
- 12. Illustrate the principle of solar water heater. What type of materials is used for that?

 $(12 \times 1 = 12 \text{ weightage})$

Part B

Answer any eight questions.

Each question carries 2 weightage.

- 13. Explain briefly the use of nano devices in sensing applications.
- 14. Discuss briefly the top-down approach in nanosynthesis.

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- 15. Do you agree with the statement 'green chemistry helps pollution control'. To what extend?
- 16. Differentiate between STO and GTO.
- 17. Explain in detail the advantages and limitations of using semi empirical and ab initio meth_{0d_3}
- 18. Outline the synthesis, host guest interaction and application of crown ethers.
- Exemplify supramolecular devices used for transport processes.
- 20. List general methods of drug synthesis.
- 21. Illustrate a radio immunoassay with example.
- Discuss briefly about combinatorial solid phase synthesis.
- 23. How can you find out surface area and porosity distribution of a solid catalyst using physical adsorption method?
- 24. Explain the fabrication and working of any one CdS based solar cell.

 $(8 \times 2 = 16 \text{ weights})$

Part C

Answer any two questions. Each question carries 4 weightage.

- 25. How can you compare the green aldol condensation and Grignard reaction with classical counter reactions?
- 26. Explain in detail the classification and nomenclature of basis sets. Give an account of the relation connecting basis set and (a) Computing time; and (b) Accuracy of a calculation.
- 27. Discuss the importance of hydrogen bonding in supramolecular chemistry.
- 28. Exemplify the industrial applications of heterogeneous catalysis.

 $(2 \times 4 = 8 \text{ weight})$