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# THIRD SEMESTER M.Sc. DEGREE (REGULAR/SUPPLEMENTARY) EXAMINATION, NOVEMBER 2023

(CBCSS)

Chemistry

## CHE 3C 10—ORGANOMETALLIC AND BIO-INORGANIC CHEMISTRY

(2019 Admission onwards)

Time: Three Hours

Maximum: 30 Weightage

#### Section A

Answer any **eight** questions. Each question carries a weightage of 1.

- Show that [V (CO)<sub>7</sub>]+ obey 18 electron rule.
- 2. What do you mean by fluxionality? Why NMR technique is preferred to other special techniques for studying fluxionality?
- 3. Internuclear spacing is the major criterion for metal metal bonding. Justify.
- 4. Give examples of compounds with metal-metal quadrupole bonds.
- 5. Illustrate Insertion reaction in organometallics.
- 6. Apply Wades rule to predict the structure of  $[Os_6(CO)_{18}]^{2-}$ .
- 7. Wacker process is carried out in the presence of of Cu (II) salt. Why?
- 8. What is meant by red drop in photosynthesis?
- 9. What is meant by co-operativity in hemoglobin?
- Give a short note on siderophores.

 $(8 \times 1 = 8 \text{ weightage})$ 

#### Section B

Answer any six questions.

Each question carries a weight of 2.

- 11. State and explain 16 electron rule as applied to organometallic compounds.
- 12. Discuss the structure of organometallic compound with allyl ligand.

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- 13. Explain water gas shift reaction.
- 14. Hydroformylation of alkene can be carried out using Cobalt complex or Rhodium complex as catal Which is more advantageous? Justify your answer.
- Discuss the role of Calcium in blood clotting.
- Sketch the isolobality of CH<sup>2-</sup> and NH<sub>2</sub>.
- Discuss the structural features and activity of tyrosinase.
- 18. Discuss the significance and mechanism of Sodium/Potassium pump.

 $(6 \times 2 = 12 \text{ weight})$ 

### Section C

Answer two questions. Each question carries a weight of 5.

- 19. Give an account of synthesis, structure and bonding in metal carbonyls.
- 20. Brief the classification of metal clusters.
- 21. Discuss the role of alkali metal ions in biological systems.
- 22. Discuss the structure and biological functions of:
  - (a) Vitamin B<sub>12</sub>.
  - (b) Catalase.

 $(2 \times 5 = 10 \text{ weight})$