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THIRD SEMESTER M.Sc. DEGREE (REGULAR/SUPPLEMENTARY) EXAMINATION, NOVEMBER 2021

(CBCSS)

Chemistry

CHE 3C 09-MOLECULAR SPECTROSCOPY

(2019 Admission onwards)

Time : Three Hours

Maximum: 30 Weightage

General Instructions

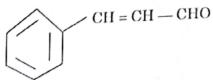
- 1. In cases where choices are provided, students can attend all questions in each section.
- 2. The minimum number of questions to be attended from the Section/Part shall remain the same.
- 3. The instruction if any, to attend a minimum number of questions from each sub section sub part! sub division may be ignored.
- 4. There will be an overall ceiling for each Section / Part that is equivalent to the maximum weightage of the Section / Part.

Section A

Answer any eight questions. Each question carries a weightage of 1.

- 1. Which of the following molecules are microwave active? Justify your answer CH2Cl2, CHCl3, CCl₄, BCl₃?
- 2. Define normal mode of vibration.
- 3. A set of protons absorb at a frequency 900 Hz higher w.r.t. TMS in 100 MHz NMR instrument. Calculate the chemical shift $\delta(\text{delta})$.
- 4. Define of factor. Explain its significance.

Predict λ_{max} for



- 6. What do you mean by first order NMR spectrum?
- 7. $\delta(\text{delta})$ values of ¹³C are appreciably higher compound to proton. Why ?

Turn over

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- 8. State and explain nitrogen rule.
- 9. What is Karplus relationship?
- 9. What is Rulpines do you expect in the EPR spectrum of benzene negative ion? Justify your and the EPR spectrum of benzene negative ion? Justify your and the EPR spectrum of benzene negative ion? Justify your and the EPR spectrum of benzene negative ion? Justify your and the EPR spectrum of benzene negative ion?

Section B

Answer any six questions. Each question carries a weightage of 2.

- 11. How would you find dipole moment of a molecule from microwave spectroscopy? Explain
- 12. Write Morse equation-represent graphically. Show that it approximates to simple harmonical for low amplitude vibrations.
- 13. How you determine spin-spin relaxation time T_2 using pulsed NMR? Discuss.
- 14. Briefly discuss zero field splitting.
- 15. How would you establish conformation and configuration of 3-methyl cyclohexanone by
- 16. Discuss the various mechanisms of spin-sprin coupling.
- 17. What is NOE? Discuss.
- 18. With the help of suitable examples discuss McLafferty rearrangement.

 $(6 \times 2 = 12 \text{ W})$

Section C

Answer any two questions. Each question carries a weightage of 5.

- 19. What are the drawbacks of conventional NMR spectrum? How are they overcome in: Discuss.
- 20. Discuss theory and applications of Mössbauer spectroscopy.
- 21. The spectral data of a compound is given IR $1620 \text{ cm}^-\text{(m)}$, $1695 \text{ cm}^{-1}\text{(s)}$

NMR: 1.9δ (3H, singlet) 2.1δ (6H, singlet)

: λ_{max} 238 nm (ϵ = 11700)

MS $\frac{m}{z}$ (100), 83 (90), 43 (78), 98 (49), 29 (46), 39 (43), 27 (42), 53 (13), 41 (13),

Deduce the structure and assign the peaks.

22. Discuss the applications of ORD in structural investigation.

(2 × 5 = 11)