

C 4715

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Name.....

Reg. No.....

**SECOND SEMESTER M.Sc. DEGREE (REGULAR/SUPPLEMENTARY)
EXAMINATION, APRIL 2021**

(CBCSS)

Chemistry

**CHE 2C 08—ELECTRO CHEMISTRY, SOLID STATE CHEMISTRY
AND STATISTICAL THERMODYNAMICS**

(2019 Admissions)

Time : Three Hours

Maximum : 30 Weightage

General Instructions

1. In cases where choices are provided, students can attend **all** questions in each section.
2. The minimum number of questions to be attended from the Section / Part shall remain the same.
3. There will be an overall ceiling for each Section / Part that is equivalent to the maximum weightage of the Section / Part.

Section AAnswer any **eight** questions.

Each question carries a weightage of 1.

1. Write electrode reactions for $H_2 - O_2$ fuel cell under alkaline conditions.
2. Find electrode potential for a calomel electrode with 0.1 M KCl. The standard electrode potential is 0.268 V, $T = 298$ K.
3. What is Stern model of electrical double layer ?
4. What is half wave potential ? Explain its significance.
5. Write Schoenflies symbol for (a) 222 ; (b) mmm.
6. Account for the semiconductivity of nonstoichiometric ZnO .
7. Explain ferrimagnetism. Write one example.
8. Explain thermodynamic probability. How is it related to entropy ?
9. Find symmetry number for (a) C_6H_6 ; (b) CH_4 .
10. Calculate the heat capacity for diamond at 1.86 K characteristic temperature is 1860 K.
($8 \times 1 = 8$ weightage)

Turn over

Section B

*Answer any six questions.
Each question carries a weightage of 2.*

11. Calculate the concentration of Ag^+ at equilibrium, when excess of finely divided metal is added to 0.05 molal ferric nitrate. The standard electrode potentials of Ag^+/Ag and $\text{Fe}^{3+}/\text{Fe}^{2+}$ are 0.799 and 0.771 V respectively. $T = 298 \text{ K}$.
12. Calculate the thickness of ion atmosphere around K^+ in 0.01 KCl at 25°C in water. T_D constant = 78.5.
13. Discuss one of the theories of hydrogen over voltage.
14. List the seven crystal systems and corresponding Bravais lattices. Discuss.
15. What is Piezoelectricity? Discuss its applications.
16. Define partition function. Derive equation to show its relationship with internal energy.
17. Calculate absolute entropy of He at 0°C and 1 atmosphere pressure.
18. Show that all particles obey Maxwell-Boltzmann statistics under dilute system conditions.

(6 × 2 = 12)

Section C

*Answer any two questions.
Each question carries a weightage of 5.*

19. What are the assumptions of Debye-Hückel theory? Using the theory derive Debye-Hückel law.
20. Derive Butler-Volmer equation.
21. Briefly discuss free electron theory of metals.
22. Discuss Debye's theory of heat capacity of solids.

(2 × 5 = 10)