

102128

(Pages : 2)

Name.....

Reg. No.....

**SECOND SEMESTER M.Sc. (CBCSS) REGULAR/SUPPLEMENTARY DEGREE  
EXAMINATION, APRIL 2024**

Chemistry

**CHE 2C 08—ELECTRO CHEMISTRY, SOLID STATE CHEMISTRY AND STATISTICAL  
THERMODYNAMICS**

(2019 Admission onwards)

Time : Three Hours

Maximum Weightage : 30

**Section A***Answer any eight questions.**Each question carries weightage 1.*

1. Write electrode reactions in methanol fuel cell.
2. Devise an electrochemical reaction in which the reaction  $\text{AgBr}_{(s)} \rightarrow \text{Ag}^+ + \text{Br}^-$  is taking place.
3. Write Helmholtz model of electrical double layer. What are its drawbacks ?
4. Explain the term 'electrode polarization'.
5. Write Hermann-Mauguin symbol for (a)  $\text{C}_{2h}$  ; (b)  $\text{D}_{2d}$ .
6. A plane cuts the  $x$ ,  $y$  and  $z$  axes at  $3a$ ,  $2b$  and  $1c$ . Write the corresponding Miller indices.
7. Explain with example 'color centers' in solids.
8. Explain the term 'most probable distribution'. How would you identify it ?
9. Write symmetry number for (a)  $\text{CH}_4$  ; (b) Ethylene.
10. State and explain equipartition principle.

 $(8 \times 1 = 8 \text{ weightage})$ **Section B***Answer any six questions.**Each question carries weightage 2.*

11. Write Debye Hückel limiting law. How is it verified experimentally ?
12. The EMF of the cell  $\text{Pt} \left| \text{H}_2 \right| \text{HCl} \left| \text{AgCl}_{(s)} \right| \text{Ag}$  is 0.3524 V at  $25^\circ\text{C}$ . The standard electrode potential of  $\text{Cl}^- \left| \text{AgCl}_{(s)} \right| \text{Ag}$  is 0.2224 V. Calculate the mean ionic activity coefficient of 0.1 molal HCl.

Turn over

13. Briefly discuss one of the theories of hydrogen over voltage.
14. Draw stereographic projection for a triclinic system. Discuss.
15. Briefly discuss working of a two stage laser.
16. Derive equations to show the relationship between partition function and (a) U
17. Calculate the heat capacity of diamond at 1000 K. Its characteristic temperature is
18. Derive Fermi Dirac distribution law.

### Section C

(6 × 2 = 12)

*Answer any two questions.  
Each question carries weightage 5.*

19. Derive Butler-Volmer equation. Discuss.
20. Briefly discuss Bore-Einstein condensation.
21. Briefly discuss band theory of solids.
22. What are the drawbacks of Einstein's theory of heat capacity of solids? How are they corrected by Debye? Discuss.

(2 × 5 = 10)